# SECONDARY ELECTRON TRANSFER IN CHROMATOPHORES OF RHODOPSEUDOMONAS CAPSULATA Ala pho<sup>+</sup>

Binary out-of-phase oscillations in ubisemiquinone formation and cytochrome  $b_{50}$  reduction with consecutive light flashes

John R. BOWYER, Gerard V. TIERNEY and Antony R. CROFTS

Department of Physiology and Biophysics, 524 Burrill Hall, University of Illinois, Urbana, IL 61801, USA

Received 22 February 1979

#### 1. Introduction

It is now well established that in isolated photochemical reaction centres of *Rhodopseudomonas* sphaeroides strain R26, a gating mechanism operates in the reduction of exogenous ubiquinone (a 2 electron carrier) by the primary acceptor complex (a 1 electron carrier) [1,2].

In chromatophores the situation is less well defined. An oscillation in proton uptake with flash number (1–2 H<sup>+</sup>/reaction centre taken up on even numbered flashes, none on odd numbered flashes) was observed [3] in *Rps. sphaeroides* carotenoidless strain R26 at pH 7.5 but not in strain Ga which contains carotenoid. In *Rhodospirillum rubrum* chromatophores, no oscillation in proton uptake with flash number was observed [4], but proton release to a carbonylcyanide *p*-trifluoromethoxyphenylhydrazone (FCCP) accessible phase oscillated with a periodicity of 2. Binary oscillations in the semiquinone signal and in the slow phase of the electrochromic shift of the carotenoid spectrum with flash number have been observed in chromatophores of

Abbreviations: DAD, 2,3,5,6,-tetramethyl-p-phenylenediamine; TMPD, N,N,N',N'-tetramethyl-p-phenylenediamine; MOPS, morpholinopropane sulphonate; cyt  $b_{50}$ , cytochrome  $b_{50}$ ; cyt  $c_2$ , cytochrome  $c_2$ ; P-870, reaction centre bacteriochlorophyll dimer; UQ, ubiquinone 10; UQ $^-$ , ubisemiquinone 10 anion;  $E_{h}(7.0)$ , ambient reduction oxidation potential at pH 7.0;  $\epsilon^{\text{red}}$ -ox, extinction coefficient difference between reduced and oxidised forms

Rps. sphaeroides w.t. and in the neurosporene containing GIC mutant but only at redox potentials of  $\geq$ 300 mV and in the presence of 50  $\mu$ M TMPD [5]. In addition, binary oscillations in the semiguinone signal in cells of Rhp. rubrum were observed in aerobic cells but not in anaerobic cells. We have suggested that cytochrome  $b_{50}$  is reduced by electrons delivered from the primary acceptor complex, either directly by  $Q_{\Pi}$  of the acceptor complex or by way of a component of equivalent stoichiometry, but without the interaction of the large pool of ubiquinone which can act as a secondary acceptor pool during multiple flash excitation in the presence of excess donor to the reaction centre [6,7]. We report here binary oscillations in cyt  $b_{50}$  reduction with flash number under conditions in which binary but out-of-phase oscillations in semiquinone formation are also observed.

### 2. Methods

Rhodopseudomonas capsulata, carotenoidless mutant A1a pho<sup>+</sup> was grown anaerobically in the medium described [8] at 29°C in 11 Roux bottles placed about 10 cm from 40 W fluorescent strip lights in a temperature-controlled cabinet with forced air circulation. Cells were subcultured 3 times before being subcultured into the 11 bottles, and they were harvested after ~24 h growth. Chromatophores were prepared by one passage through a

French Pressure Cell at 13 000–14 000 1b/in<sup>2</sup>. Unbroken cells and large fragments were removed by centrifugation for 20 min at 17 000  $\times$  g and for 45 min at 35 000  $\times$  g.

For kinetic measurements flash excitation was provided by a Xe flash, 90–95% saturating, of  $\sim$ 5  $\mu$ s pulse width, and absorbance changes recorded using a rapidly responding single beam kinetic spectrophotometer [9]. Transients were averaged after being read into a PDP 11/34 computer (Digital Equipment Corp.) via a Datalab DL905 Transient Recorder (Data Labs, Mitcham, Surrey, CR4 4 HR). The spectrophotometer was equipped with a shutter so that chromatophores were only exposed to the measuring light for the few seconds during which absorbance changes were recorded. For redox titrations, a stirred anaerobic redox cuvette was used [10], but for recording spectra of the flash number dependency of excitation-induced absorbance changes a redox poised reservoir in an anaerobic flow system was used to provide dark-adapted material. The whole system was computer controlled through a DEC Laboratory Peripheral System 11 using an appropriate assembly language program.

Chromatophores were routinely suspended in 50 mM MOPS, 100 mM KCl (pH 7.0), with valino-mycin added to 2  $\mu$ M to prevent build up of a membrane potential.

An  $e^{\text{red}-\text{ox}}$  of 19.5 mM<sup>-1</sup> cm<sup>-1</sup> at 605 nm, calculated from [11] was used to estimate the extent of *P*-870 photo-oxidation. This value was determined in [11] for *Rps. sphaeroides* Ga. No value is available for *P*-870 in chromatophores in *Rps. capsulata* A1a pho<sup>+</sup>; however our results suggest 18.5  $\pm$  3.6 mM<sup>-1</sup> cm<sup>-1</sup>.

A minimum of 2 min dark time was allowed between measurements. Increasing the dark time did not significantly affect the oscillatory patterns observed except at the lower end of the redox potential range under study ( $\leq 300 \text{ mV}$ ).

Cytochrome  $b_{50}$  kinetics were recorded at 560—570 nm, a wavelength pair which almost entirely eliminates  $\Delta A$  due to P-870 and cyt  $c_2$  in Rps. capsulata A1a pho<sup>+</sup> (J.R.B., unpublished observations). Since most measurements of oscillations were made at 370—400 mV, where mediators such as TMPD and DAD had little effect on the P-870 rereduction kinetics in the presence of 0.5—1.0 mM potassium

ferrocyanide (result not shown), potassium ferrocyanide was the only exogenous electron donor present except where indicated. This also avoided complications arising from rapid  $\Delta A$  due to mediator oxidation and reduction encountered [5]. Under these conditions P-870 rereduction was extremely slow (85% complete after 10 s at  $E_{\rm h(7.0)}$  450 mV rising to ~100% complete after 5 s at  $E_{\rm h(7.0)}$   $\leq$ 370 mV. In order that significant reaction centre rereduction and cyt  $b_{50}$  reoxidation could occur between flashes, the time between each flash in a train was 5–10 s. Although some P-870 rereduction by the return of the electron from the acceptor complex may have occurred over this time scale, it was not sufficient to obscure oscillations.

#### 3. Results and discussions

## 3.1. Binary oscillations in cytochrome b<sub>50</sub> kinetics

The traces of fig.1 show the cyt  $b_{50}$  reduction kinetics in a train of 4 flashes at  $E_{h(7,0)}$  370 mV and  $E_{h(7,0)}$ 300 mV. Both the extent and initial rate of cyt  $b_{50}$ reduction showed a binary oscillation with very little cyt  $b_{50}$  being reduced on flash 1 and 3 at  $E_{h(7.0)}$ 370 mV, and considerably more going reduced on flashes 2 and 4. The oscillation had almost disappeared at 300 mV, due to an increase in the extent of cyt  $b_{50}$  reduction on flash 1 and a decrease in that on flash 2. Detailed spectra of the flash-induced change on each flash in a train of 4 flashes from the darkadapted state were recorded in the spectral region 530–580 nm where the  $\alpha$  band of cyt  $b_{50}$  is located. After correction for the reaction centre change, they showed that the component(s) going reduced on flashes 2, 3 and 4 had identical spectral characteristics.

In the absence of antimycin A the extent of cyt  $b_{50}$  reduction on flashes 2, 3 and 4 was considerably reduced, presumably due to a rapid antimycinsensitive reoxidation process, but fig.3A,B shows that oscillations could still be observed.

## 3.2. Binary oscillations in the redox state of UQ

At least 4 redox components can show flash number-dependent  $\Delta A_{420-480}$  where ubisemiquinone redox changes are recorded. These are *P*-870, UQ, cyt  $b_{50}$  ( $\gamma$  band) and cyt  $c_2$  ( $\gamma$  band). At 446 nm, close to the peak of the UQ'--UQ difference

spectrum in methanol [12],  $\Delta A$  due only to P-870 and UQ should be observed. Since there was no flash number-dependent variation in the extent of P-870 photo-oxidation at <390 mV (fig.1) oscillations in the  $\Delta A_{446}$  should reflect the redox state of UQ. The

results shown in fig.1 suggest that UQ $^{-}$  was formed on odd numbered flashes and disappeared on even numbered flashes, i.e., out of phase with cyt  $b_{50}$  reduction. This oscillation showed a similar redox potential dependency as the cyt  $b_{50}$  oscillation, i.e.,

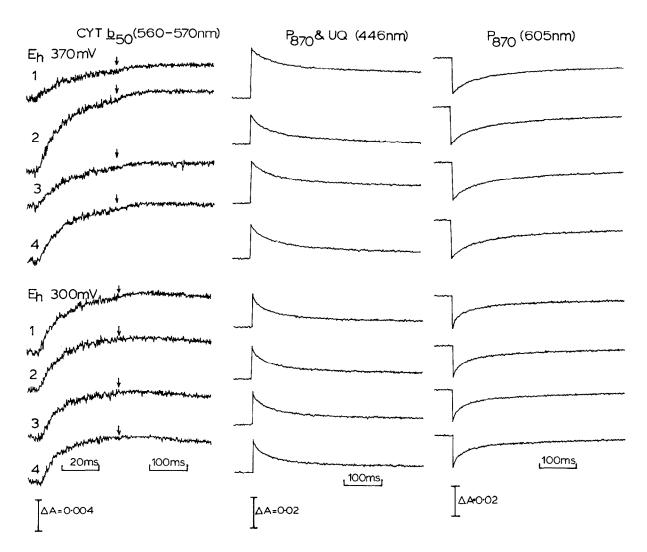


Fig. 1. Flash number dependency of  $\Delta A_{560-570}$ ,  $\Delta A_{446}$  and  $\Delta A_{605}$ . For the cyt  $b_{50}$  measurements, chromatophores were suspended to ~1.4  $\mu$ M reaction centres in 50 mM MOPS, 100 mM KCl (pH 7.0) containing a total of 6 mM potassium ferrocyanide, 6 mM potassium ferricyanide, 10  $\mu$ M TMPD, 2  $\mu$ M valinomycin and 2  $\mu$ M antimycin A, in an anaerobic redox cuvette. Signals were not averaged. The instrument response time was 200  $\mu$ s. The time between flashes in a group was 5 s. The dark time between groups was 2 min. The vertical arrows indicate where the transient recorder time base was changed. For the 446 nm and P-870 measurements, chromatophores were suspended to ~1.4  $\mu$ M reaction centres in 50 mM MOPS, 100 mM KCl (pH 7.0) containing a total of 2 mM potassium ferrocyanide and ferricyanide, 2  $\mu$ M valinomycin and 2  $\mu$ M antimycin A in an anaerobic redox cuvette. Signals were not averaged. The instrument response time was 500  $\mu$ s. The time between flashes in a group was 10 s. The dark time between groups was 2 min.

binary oscillations were marked at 370 mV but they gradually damped as the potential was lowered to 300 mV. Antimycin A had no effect on the UQ oscillation. The difference between the rapid kinetics of the  $\Delta A_{446}$  on flash 1 and flash 2 should reflect the rate of conversion of UQ to ubiquinol (fig.2). This process had a  $t_{1/2}$  30–50  $\mu$ s, considerably faster than the rates observed for  $Q_{II}^{-}$  to  $Q_{II}^{-}$  transfer in Rps. sphaeroides R26 reaction centres [13,14].

## 3.3. Oscillatory $\Delta A_{420-480}$

Figure 3A–D show spectra of the flash-induced  $\Delta A$  at various times after the flash for flashes 1 and 2 from the dark-adapted state in the absence and presence of antimycin A. The band peaking at 430 nm had the kinetics of cyt  $b_{50}$  (traces not shown). In resolving the spectra it was assumed that the  $\Delta A$  recorded 1 ms after the flash on flash 2 in the absence of antimycin A would be largely due to P-870, since the spectra were recorded at  $E_{h(7.0)}$  370–390 mV where the extent of rapid cyt  $c_2$  oxidation in this strain is small [10]. From these data a set of normalising factors was calculated to correct other spectra for the P-870 change using 540 nm as a reference wave-

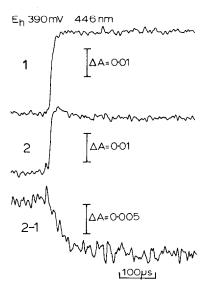


Fig. 2. Rapid kinetics of the change recorded at 446 nm on the first 2 excitations from the dark-adapted state, and the difference kinetics of the 2 absorption changes. The traces were obtained using the same set up as in fig. 3C, D. Traces (32) are averaged. The instrument response time was 1 µs.

length at which only *P*-870 changes would be observed. Figure 3EF, show the spectra of the changes recorded on flash 1 and 2 in the presence of antimycin A, at 1 ms and 50 ms, respectively, after the flash, cor-

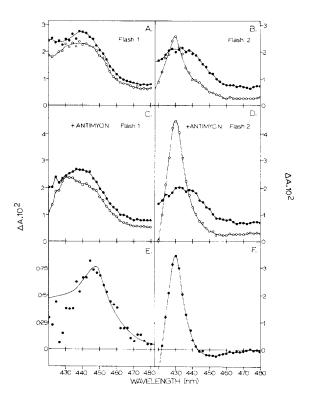


Fig.3. (A-D) Flash number dependency of the spectra of the excitation-induced  $\Delta A_{420-480}$  from the dark-adapted state. (A,B) Chromatophores were suspended  $\sim 1.4 \mu M$  reaction centres in 50 mM MOPS, 100 mM KCl (pH 7.0) containing a total of 1 mM potassium ferricyanide, 1 mM ferrocyanide and 2 µM valinomycin in a 25 ml total vol. in an anerobic flow system. The  $E_{h(7.0)}$  was 390 mV. The time between flashes in a group was 10 s. The spectra show changes recorded 1 ms (•) and 50 ms (o) after both flash 1(A) and flash 2(B). Signals were not averaged. The instrument response time was 200  $\mu$ s. (C,D) As for (A,B) except that antimycin A was present at 2 μM. (E) The points (•) show the excitation-induced  $\Delta 4$  1 ms after flash 1 in a train in the presence of antimycin A (as in fig.3C) corrected for the reaction centre changes as described in the text, using the change recorded at 540 nm as a measure of the reaction centre change alone. Superimposed is the difference spectrum for UQ'--UQ in methanol (from [12]). (F) Spectrum of the flash-induced  $\Delta A$  50 ms after flash 2 in a train in the presence of antimycin A (as in fig.3D) corrected for the reaction centre change as above and in the text.

rected for P-870 changes. Figure 3F represents the difference spectrum of cyt  $b_{50}$ . Superimposed on fig.3E is the difference spectrum for UQ' $^-$ UQ in methanol [12]. The fit is good at >440 nm, but there appears to be some additional spectral complexity at 420–440 nm.

#### 3.4. Oscillations in the rate of P-870 rereduction

The rate of P-870 rereduction also showed binary oscillations, being more rapid after even numbered flashes than after odd numbered flashes. This oscillation was most clearly evident in the absence of antimycin A, and presumably again reflects the operation of a 2 electron gate, and its effect on the availability of electrons to the chain reducing P-870.

# 3.5. Correlation of the oscillations in $\Delta A_{446}$ and $\Delta A_{560-570}$

It seems likely that the oscillations in  $\Delta A$  reflect a 2 electron gate acting between the primary acceptor complex and cyt  $b_{50}$ . This may be described by the following scheme, which ignores protonation events:

$$\begin{aligned} & Q_{\mathbf{I}}Q_{\mathbf{II}}b_{50}^{\mathrm{ox}} \cdot \mathbf{A}^{\mathrm{ox}} \xrightarrow{h\nu_{1}} Q_{\mathbf{II}}b_{50}^{\mathrm{ox}} \cdot \mathbf{A}^{\mathrm{ox}} \xrightarrow{\operatorname{dark}} \\ & Q_{\mathbf{I}}Q_{\mathbf{II}}^{-1}b_{50}^{\mathrm{ox}} \cdot \mathbf{A}^{\mathrm{ox}} & \longrightarrow \\ & Q_{\mathbf{I}}Q_{\mathbf{II}}^{-1}b_{50}^{\mathrm{ox}} \cdot \mathbf{A}^{\mathrm{ox}} \xrightarrow{h\nu_{2}} Q_{\mathbf{I}}^{-1}Q_{\mathbf{II}}^{-1}b_{50}^{\mathrm{ox}} \cdot \mathbf{A}^{\mathrm{ox}} \xrightarrow{\operatorname{dark}} \\ & Q_{\mathbf{I}}Q_{\mathbf{II}}^{2}b_{50}^{\mathrm{ox}} \cdot \mathbf{A}^{\mathrm{ox}} \xrightarrow{\rightarrow} Q_{\mathbf{I}}Q_{\mathbf{II}}b_{50}^{\mathrm{red}}\mathbf{A}^{\mathrm{red}} \end{aligned}$$

where A is an unidentified acceptor.

In chromatophores of Rps. capsulata A1a pho<sup>+</sup>, binary oscillations occurred for  $\gg$ 4 flashes, suggesting that  $Q_{\Pi}H_2$  (the presumed fully protonated ubiquinol) is returned to its initial state after the second flash, i.e., is oxidised to  $Q_{\Pi}$ , giving up 2 electrons and 2 H<sup>+</sup> in the process. The reduction of cyt  $b_{50}$  is a 1 hydrogen transfer event below its pK, and the fate of  $Q_{\Pi}^{-}$  (or  $Q_{\Pi}H$ ) generated by the proposed reduction of cyt  $b_{50}$  by  $Q_{\Pi}H_2$  is uncertain. We might consider the possibility that the extra electron reduces another molecule of cyt  $b_{50}$ , or alternatively that some other component, possibly J [15,16] or UQ acts as an acceptor in a parallel reaction. In the latter cases we could not predict whether cyt  $b_{50}$  reacts with the quinol/semiquinone couple or the quinone/semi-

quinone couple. Using an extinction coefficient for cyt  $b_{50}$  at 560-570 nm of 13.2 mM<sup>-1</sup> cm<sup>-1</sup> (determined experimentally for isolated beef heart mitochondrial cyt b [17] and calculated from the data in [18,19]) we determined that  $\leq$ 0.6 molecules cyt  $b_{50}/P$ -870 oxidised were reduced in the presence of antimycin A on flash 2 at  $E_{h(7.0)}$  370 mV. The possibility that more cyt  $b_{50}$  was reduced on flash 2 but was rapidly reoxidised in an antimycin A-insensitive fashion cannot, of course, be excluded.

The pattern of cyt  $b_{50}$  reduction kinetics indicates some damping in the primary acceptor complex. This is also evident in the oscillations at 446 nm, but analysis of damping at this wavelength is complicated by the occurrence of overlapping reaction centre changes. Using  $\epsilon = 8.5 \text{ mM}^{-1} \text{ cm}^{-1}$  for UQ<sup>--</sup>–UQ obtained with reaction centres of *Rps. sphaeroides* R26 [20] we calculated that ~1 UQ<sup>--</sup> was formed/reaction centre oxidised on flash 1. However, this is likely to be an overestimate, since some cyt  $b_{50}$  reduction occurred on flash 1 indicating that some centres had  $Q_{\Pi}$  reduced before flash 1, so it may be that the above  $\epsilon$  is too low.

# 3.6. The diminution in oscillations with falling redox potential

A number of factors may have contributed to the apparent disappearance of oscillations as the redox potential was lowered:

(i) A change in the mechanism of electron flow from the primary acceptor complex to cyt  $b_{50}$  such that charge accumulation no longer occurs, e.g., direct reduction of cyt  $b_{50}$  by  $Q_{I}^{-}$  (or  $Q_{I}H$ ) or by  $Q_{II}^{-}$  (or  $Q_{II}H$ ).

This may result from the presence before the flash of some reduced component outside the reaction centre complex, e.g., cyt  $c_2$  or J. We intend to test whether cyt  $c_2$  is involved using a mutant lacking cyt  $c_2$  isolated by B. Marrs. The apparent non-chromophoric proton binding agent in isolated reaction centres of Rps. sphaeroides R26 appears to have a redox potential-dependent pK [13] and this might affect the gate mechanism in chromatophores.

(ii) It now appears that with cyt  $c_2$  reduced, ~30% of the reaction centres in chromatophores may be rereduced with  $t_{1/2} < 5 \mu s$  following laser flash excitation (R. E. Overfield and C. A. Wraight, personal communication; J.R.B. unpublished

- observations) and since the rate of electron transfer between  $Q_{\rm I}$  and  $Q_{\rm II}$  in chromatophores appeared to have  $t_{1/2}$  30–50  $\mu{\rm s}$ , it is possible that the Xe flash caused some double hits as cyt  $c_2$  became reduced before the flash.
- (iii) As the redox potential is lowered the measuring beam would become more actinic because the presence of reduced cyt  $c_2$  would allow a more rapid turnover of the photochemistry. However, in these experiments, the brief exposure to the measuring beam (10–30 pEinstein s<sup>-1</sup> cm<sup>-2</sup>) before flash excitation would make this effect insignificant.

### Acknowledgements

We are indebted to Colin Wraight for his very valuable comments and access to his unpublished results. J.R.B. thanks the Research Board of the University of Illinois for financial support. We are grateful to Miss Beth Ransdell for assistance in growing the bacteria.

#### References

- [1] Wraight, C. A. (1977) Biochim. Biophys. Acta 459, 525-531.
- [2] Vermeglio, A. (1977) Biochim. Biophys. Acta 459, 516-524.
- [3] Barouch, Y. and Clayton, R. K. (1977) Biochim. Biophys. Acta 462, 785-788.

- [4] Fowler, C. F. (1976) Int. Conf. Primary Electron Transport and Energy Transduction in Photosynthetic Bacteria, Brussels (Dr C. Sybesma, org.) abstr. WB9.
- [5] De Grooth, B. G., Van Grondelle, R., Romijn, J. C. and Pulles, M. P. J. (1978) Biochim. Biophys. Acta 503, 480–490.
- [6] Crofts, A. R. and Bowyer, J. R. (1978) in: The Proton and Calcium Pumps (Azzone, G. F. et al. eds) pp. 55-64, Elsevier/North-Holland Biomedical Press, Amsterdam, New York.
- [7] Bowyer, J. R., Baccarini Melandri, A., Melandri, B. A. and Crofts, A. R. (1978) Z. Naturforsch, 33c, 704-711.
- [8] Sistrom, W. R. (1960) J. Gen. Microbiol. 22, 778-785.
- [9] Jackson, J. B. and Crofts, A. R. (1971) Eur. J. Biochem. 18, 120-130.
- [10] Evans, E. H. and Crofts, A. R. (1974) Biochim. Biophys. Acta 357, 89-102.
- [11] Dutton, P. L., Petty, K. M., Bonner, H. S. and Morse, S. D. (1975) Biochim. Biophys. Acta 387, 536-556.
- [12] Bensasson, R. and Land, E. J. (1973) Biochim. Biophys. Acta 325, 175-181.
- [13] Wraight, C. A. (1979) Biochim. Biophys. Acta, submitted.
- [14] Vermeglio, A. and Clayton, R. K. (1977) Biochim. Biophys Acta 461, 159–165.
- [15] Bowyer, J. R. and Crofts, A. R. (1978) in: Frontiers of Biological Energetics (Dutton, P. L. et al. eds) vol. 1, pp. 326-333, Academic Press, London, New York.
- [16] Crofts, A. R. (1979) in: Light-Induced Charge Separation in Biology and Chemistry (Gerischer, H. and Katz, J. J. eds) Dahlem Konferenzen, Berlin, in press.
- [17] Goldberger, R., Smith, A. L., Tisdale, H. and Bomstein, R. (1961) J. Biol. Chem. 236, 2788-2793.
- [18] Erecińska, M., Wilson, D. F. and Mujata, Y. (1976) Arch. Biochem. Biophys. 177, 133–143.
- [19] Arutjunjan, A. M., Kamensky, Y. A., Milgröm, E., Surkov, S., Konstantinov, A. A. and Sharonov, Y. A. (1978) FEBS Lett. 95, 40-44.
- [20] Wraight, C. A., Cogdell, R. J. and Clayton, R. K. (1975) Biochim. Biophys. Acta 396, 242–249.
- [21] Petty, K. M. and Dutton, P. L. (1976) Arch. Biochem. Biophys. 172, 346-353.